

# Light and Spectroscopy

## Equipment

- Photometers, spectrometers, gratings
- Various light sources
- Light boxes and color filters

## Objective

Physics Concepts

- Irradiance
- Spectroscopy

Experimental analysis

- Fit curves to data to determine mathematical relationships
- Recognizing the uncertainty in measurements

## Conceptual (C-Level)

Given an ideal point source of light that radiates in all directions determine what shape you expect the irradiance as a function of radial distance to look like.

Graph the “ideal” spectrum (intensity as a function of  $\lambda$  and  $f$ ) for the following:

- A blue object
- A red object
- A yellow object

Create a Venn diagram for color addition and another for color subtraction.

### EXPLORATIONS:

Put on the diffraction glasses (also called “fireworks glasses”) and observe different light sources.

## Basic Lab (B-Level)

- Measure the irradiance as a function of distance.
  - What is a “Lux”?
  - Comment on the shape of your graph.
- Using the spectrometer, graph the spectra from three different spectral tubes [H, He, Ne, Hg].
  - Determine the expected spectra and compare with your results. NOTE: Consider referencing other sources for information on spectral lines.

**Danger High Voltage!** The spectra tubes are powered by a high voltage source. Make sure the source is unplugged before changing tubes.

**Caution Fragile!** The spectra tubes are fragile. Please handle the tubes by their ends. Please avoid getting oil (from your hands) on the tubes – especially the center thin region.

## Advanced/Extended Lab Ideas (A-Level)

- Graph the spectra from other light sources.
  - Incandescent light bulb (try dimming it), candle, fluorescent light bulb...
- Graph absorption and/or transmission spectra for different solutions.
  - Chloroplasts
  - Concentration of creamer in water. Can you get a “red sunset”?
- What are you curious to investigate?